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15 1 mol of water (18 g) contains 6 1023 molecules. Therefore, 18 cm3 of water contains 6 1023 molecules. (Density is 1 g cm3) 1 cm3 of water contains 6 1023 /18 = 3.333 1022 molecules. 0.05 cm3 of water contains 0.05 3.333 1022 = 1.67 1021 molecules. Chapter 23 1. From the equation, 4 mol Na gives 1 mol Ti.

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